

FORMATION OF VIBRATIONAL DISTRIBUTION FUNCTION OF ELECTRONICALLY-EXCITED $N_2(A^3\Sigma_u^+)$ MOLECULES IN NITROGEN DISCHARGE PLASMA

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ABSTRACT

A model on formation of the vibrational distribution function of metastable electronically excited $N_2(A^3\Sigma_u^+, \nu)$ states in nitrogen discharge plasma is presented. It is assumed that the probability of population of $N_2(A^3\Sigma_u^+, \nu)$ states by collisional deactivation of $N_2(B^3\Pi_g, \nu')$ molecules is proportional to the Franck-Condon factors of corresponding $N_2(B^3\Pi_g, \nu') \rightarrow N_2(A^3\Sigma_u^+, \nu)$ transitions, i.e. that the quenching is essentially vertical. The results of calculations carried out under this assumption are in agreement with measured $N_2(A^3\Sigma_u^+, \nu)$ vibrational distribution at the end of pulsed high current discharge in nitrogen.

1. INTRODUCTION

Electronically excited $N_2(A^3\Sigma_u^+, \nu)$ molecules play an important role in gas discharge kinetics in nitrogen and nitrogen-oxygen mixtures [1-4]. In mixtures with a small fraction of oxygen a considerable part of the discharge energy, spent on excitation of electronic degrees of freedom, accumulates in $N_2(A^3\Sigma_u^+)$ state [3]. An important feature of many reactions with $N_2(A^3\Sigma_u^+, \nu)$ molecules is a dependence of reaction rates and product ratios on a number of the vibrational level.

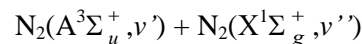
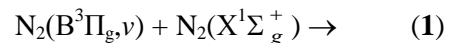
In a number of recent works, the methods of laser-induced fluorescence (LIF) [2] and intracavity laser absorption spectroscopy at the transitions of the 1^+ system of nitrogen were used for measuring the density of $N_2(A^3\Sigma_u^+)$ molecules [5]. Populations of only a few

vibrational levels of $N_2(A^3\Sigma_u^+)$ can be found using these techniques. Thus a model describing the formation and evolution of the vibrational distribution function of electronically excited $N_2(A^3\Sigma_u^+, \nu)$ molecules in nitrogen discharge plasma is needed for more detailed comparison of experimental results with numerical simulations.

2. MODEL DESCRIPTION

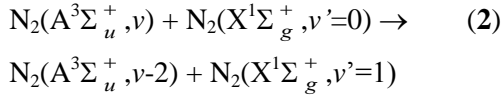
In nitrogen discharge plasma the population of $N_2(A^3\Sigma_u^+, \nu)$ state occurs mainly due to the radiative-collisional quenching of levels with high energies $N_2(B^3\Pi_g)$, $N_2(B^3\Sigma_u^-)$, $N_2(W^3\Delta_u)$ and others, the excitation by direct electron impact is not very important [3].

Analysis of experimental results on the population of vibrational levels of electronically excited N_2 states due to recombination of $N(^4S)$ atoms in the afterglow of nitrogen discharge ($P = 20 - 760$ Torr) is presented in [6]. It is shown that the most part of the excitation flux, passed through the levels $N_2(B^3\Pi_g, \nu=0-12)$, also passes through the levels $N_2(A^3\Sigma_u^+, \nu'=0,1)$. This evidences that in quenching of $N_2(B^3\Pi_g, \nu)$ molecules the reaction



is dominant. Similar conclusion based on analysis of experimental data has been made in [7]. This conclusion is really important since in several works the authors assumed the reaction $N_2(B^3\Pi_g, \nu) + N_2(X) \rightarrow N_2(X, \nu') + N_2(X, \nu'')$ as a dominant quenching channel of $N_2(B^3\Pi_g, \nu)$ state.

It is important to underscore that the question about vibrational distribution function of $N_2(A^3\Sigma_u^+, \nu)$ molecules generated in reaction (1) is still open. Calculations of the temporal evolution of $N_2(A^3\Sigma_u^+, \nu=0-8)$ vibrational levels populations in the afterglow of pulsed high current electron beam and RF discharge in nitrogen have been performed in [3]. Comparison of numerical and experimental results allowed improving the rate constants of reaction



Data obtained in [3] were used in this study. Having uniformity of all parameters in the discharge the following expression for the electric field E was used in the present paper:

$$E = \frac{I}{e\mu_e N_e S}, \quad (3)$$

where I is the discharge current, μ_e is the electron mobility (function of E/N), N_e is the electron density, S is the area of the discharge cross-section.

Dependences of the rates of excitation, ionization and dissociation of molecules by the electron impact on the reduced electric field E/N value are determined from the calculations of the electron energy distribution function (EEDF) using the BOLSIG+ code [8]. Main processes with neutral and charged particles and the rate constants are taken from [9]. Detailed description of the model was given in [3].

3. SIMULATION RESULTS

3.1. Time evolution of population difference between $N_2(A^3\Sigma_u^+, \nu=0)$ and $N_2(B^3\Pi_g, \nu=2)$ states were measured in [4] in a positive column of DC nitrogen pulsed discharge. The discharge was created in a 14 mm inner diameter, 25 cm long pyrex tube. High-voltage pulses of about 2 kV of $\tau_{\text{imp}} = 8 - 20 \mu\text{s}$ duration and 20 Hz repetition rate were applied to produce bulk discharge in pure nitrogen.

Fig. 1 shows the discharge current [4] and the calculation results of temporal evolution of the reduced electric field E/N under the conditions of [4]. It is clear that the main energy deposition

occurs at $E/N = 120 - 130 \text{ Td}$. The population difference between $N_2(B^3\Pi_g, \nu = 2)$ and $N_2(A^3\Sigma_u^+, \nu=0)$ states was evaluated from integrated high-resolution laser absorption/gain profiles. Optical gain, due to the population inversion, was present during the first 3 - 6 μs of the discharge pulse.

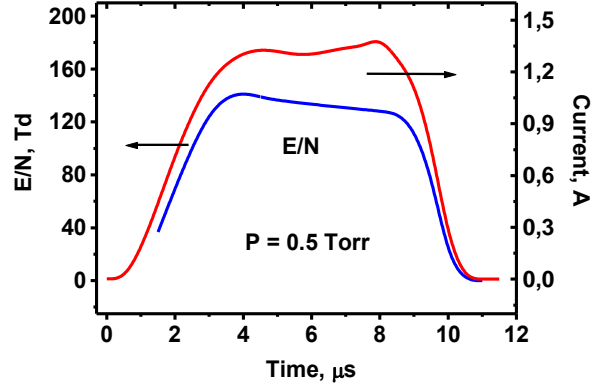


Fig. 1. Temporal evolution of the reduced electric field E/N and the discharge current under the conditions of [4]: $P = 0.5 \text{ Torr}$.

The population difference was about $\Delta N_{\text{BA}} = 3 \cdot 10^{11} \text{ cm}^{-3}$ at the peak current at 6 - 7 μs . In the afterglow the ΔN_{BA} value decreases to $\Delta N_{\text{BA}}^0 = -1.1 \cdot 10^{12} \text{ cm}^{-3}$. **Fig. 2** shows the experimental data and calculation results of the population difference between $N_2(B^3\Pi_g, \nu=2)$ and $N_2(A^3\Sigma_u^+, \nu=0)$ at $P = 0.5 \text{ Torr}$ under the conditions of Fig. 1. The main population of the $N_2(B^3\Pi_g, \nu=2)$ state in this case occurs by electron impact from the ground $N_2(X^1\Sigma_g^+)$ state and quenching reaction of electronically excited $N_2(C^3\Pi_u)$ molecules.

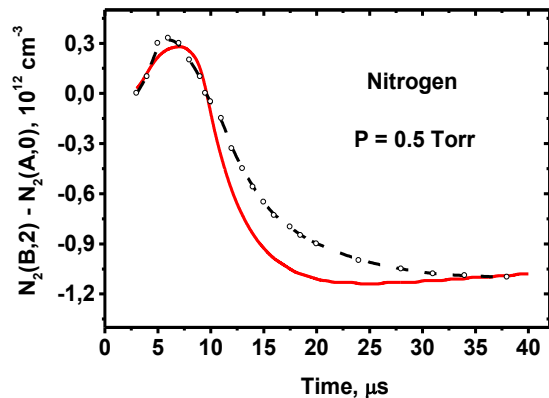


Fig. 2. Temporal evolution of number density difference: $[N_2(B^3\Pi_g, \nu=2)] - [N_2(A^3\Sigma_u, \nu=0)]$ under the conditions of [4]. The full curve represents experimental data, the dotted curve – calculation results.

The production of $N_2(A^3\Sigma_u^+, v=0)$ molecules is associated with radiative - collisional quenching of $N_2(B^3\Pi_g, v)$ state in the reaction (1). As can be seen, the calculation results are consistent with the experimental data, which indicates the adequacy of the description of the $N_2(A^3\Sigma_u^+, v=0)$ molecules production under given conditions [4].

3.2. As it was mentioned earlier at present there is no data on the vibrational distribution of $N_2(A^3\Sigma_u^+, v)$ molecules, produced at quenching of $N_2(B^3\Pi_g)$ by nitrogen molecules. In given work it is assumed that the population probability of $N_2(A^3\Sigma_u^+, v)$ in the process of collisional deactivation of $N_2(B^3\Pi_g, v')$ is proportional to the Franck - Condon factors (FCF) of corresponding $N_2(B^3\Pi_g, v') \rightarrow N_2(A^3\Sigma_u^+, v)$ transitions.

For validation of this assumption the calculation results of $N_2(A^3\Sigma_u^+, v)$ vibrational distribution have been compared with the experimental data of [5]. In [5] by means of intracavity laser absorption spectroscopy the vibrational distribution of $N_2(A^3\Sigma_u^+, v=2-6)$ states have been measured after pulsed ($\tau_{imp} = 700$ ns) high current discharge in nitrogen at $P = 230$ Torr and $T_0 = 300$ K.

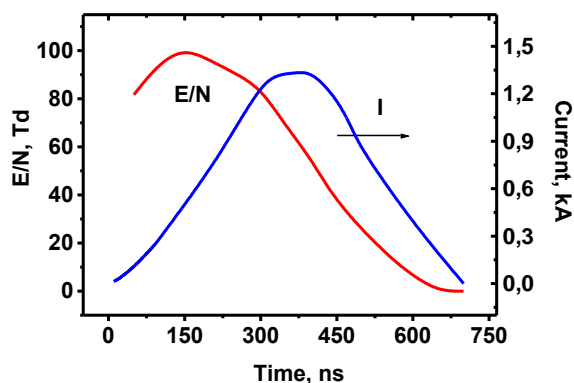


Fig. 3. Temporal evolution of the reduced electric field E/N and the discharge current in nitrogen under the conditions of [5].

The area of electrodes was 5×50 cm² with the interelectrode distance $d = 4$ cm. Having uniformity of the discharge parameters the expression (3) for the electric field E was also used under given conditions. **Fig. 3** shows the current pulse oscillogram [5] and the calculation results of temporal dynamics of the reduced electric field E/N under the conditions of [5]. It is clear that the main energy deposition occurs at

$E/N = 50 - 100$ Td. Specific energy deposited into the discharge achieved $W = 7.5 \cdot 10^{-3}$ eV/molecule, and that agrees with the experimental estimate ($7.8 \cdot 10^{-3}$ eV/molecule).

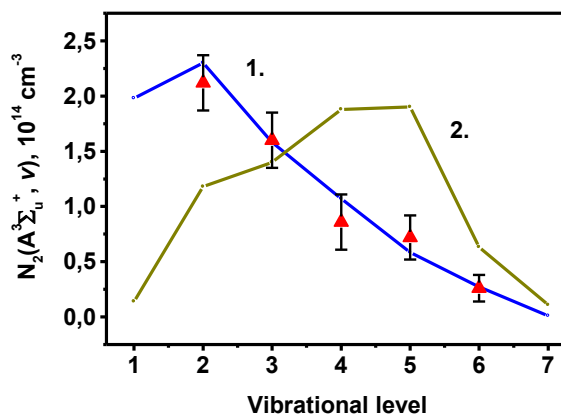
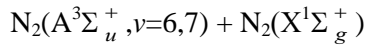
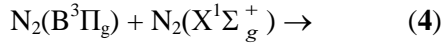


Fig. 4. Comparison between experimental [5] and predicted $N_2(A^3\Sigma_u^+, v = 1-7)$ vibrational level populations at the end of discharge at $t = 0.7$ μ s. The dots are experimental data from [5], curves (1, 2) – are calculation results (see text).

The calculation results of vibrational distribution of $N_2(A^3\Sigma_u^+, v)$ and the experimental data of [5] are shown in **fig. 4**. In this case the main excitation of $N_2(A^3\Sigma_u^+, v)$ molecules is due to the quenching of $N_2(B^3\Pi_g)$ state. The calculation results and experimental data of [5] are taken for $t = 0.7$ μ s, correspondent to the end of the current pulse. The calculations made on the assumption, that the population probability of $N_2(A^3\Sigma_u^+, v)$ in the process of collisional quenching of $N_2(B^3\Pi_g)$ was proportional to the Franck-Condon factors of the correspondent $N_2(B^3\Pi_g, v') \rightarrow N_2(A^3\Sigma_u^+, v)$ transitions, are shown with **curve 1**. A good agreement with measurements of [5] is evident. In conditions under consideration the characteristic times of reaction (2) for vibrational levels $v = 5, 4, 3$ amount to 0.44, 0.94 and 2.9 μ s, respectively. Therefore at time scales $t \leq 1$ μ s an intensive quenching of $N_2(A^3\Sigma_u^+, v \geq 4)$ vibrational levels and hence populating of levels $v = 2$ and 3 take place. And this is the main reason for relative overpopulation of the level $v = 2$ at $t = 0.7$ μ s observed in **fig. 4 (curve 1)**.

It should be noted that in a few works it is assumed that in a quenching process of $N_2(B^3\Pi_g)$ state the high vibrational levels of $N_2(A^3\Sigma_u^+, v)$ are mainly populated:



For validation of this assumption the calculations with an allowance for the quenching of $\text{N}_2(\text{B}^3\Pi_g)$ state in accordance with (4) have been carried out. **Fig. 4 (curve 2)** illustrates the results of calculations under the conditions of experiment [5]. Obtained results obviously differ from the experimental data [5]. Thus the assumption about the dominant populating of high vibrational levels of $\text{N}_2(\text{A}^3\Sigma_u^+, v \geq 6)$ owing to the collisional quenching of $\text{N}_2(\text{B}^3\Pi_g)$ state seems to be invalid.

4. CONCLUSIONS

A model on formation of vibrational distribution function of metastable electronically excited $\text{N}_2(\text{A}^3\Sigma_u^+, v)$ states in gas discharge plasma is presented. In nitrogen discharge plasma the excitation of $\text{N}_2(\text{A}^3\Sigma_u^+)$ molecules takes place mainly by deactivation of electronic states with high energies. In the paper it was assumed that the population probability of $\text{N}_2(\text{A}^3\Sigma_u^+, v)$ in the process of collisional quenching of $\text{N}_2(\text{B}^3\Pi_g)$ state is proportional to the Franck-Condon factors of correspondent transitions $\text{N}_2(\text{B}^3\Pi_g, v') \rightarrow \text{N}_2(\text{A}^3\Sigma_u^+, v)$, i.e. that the quenching is essentially vertical. The results of calculations carried out under this assumption are in agreement with measured data of $\text{N}_2(\text{A}^3\Sigma_u^+, v)$ vibrational distribution at the end of pulsed high current discharge in nitrogen at $P = 230$ Torr [5].

It is also shown that the assumption about preferred population of high vibrational levels $\text{N}_2(\text{A}^3\Sigma_u^+, v \geq 6)$ in the process of collisional quenching of $\text{N}_2(\text{B}^3\Pi_g)$ is not valid since in this case the modeling results are inconsistent with the experimental data of [5].

Time evolution of population difference between $\text{N}_2(\text{A}^3\Sigma_u^+, v=0)$ and $\text{N}_2(\text{B}^3\Pi_g, v=2)$ states in a positive column of nitrogen glow discharge was calculated under the conditions of experiments [4]. The calculation results are consistent with the experimental data, which indicates the adequacy of the description of the $\text{N}_2(\text{A}^3\Sigma_u^+, v=0)$ molecules production under given conditions.

ACKNOWLEDGMENTS

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